

ROZANOVA, V. D. and YENIKEYEVA, S. I.

The Physiological Mechanisms of Reactivity Upon the Action of Tetanus Toxin
at Different Age Periods p. 84.

ROZANOVA, V. D. and GALEYEVA, L. S. "Characteristics of Collapse in Dysentery
Intoxication and Artificial Immunity Against Dysentery p.114

Problema Reaktivnosti v Patologii, Medgiz, Moscow 1954, 344 p
(The Problem of Reactivity in Pathology)

USSR/Medicine - Physiology

FD-1332

Card 1/1 : Pub 33-10/25

Author : Rozanova, V. D.

Title : Changes in stability of heart to atropine depending on age

Periodical : Fiziol. zhur. 4, 452-457, Jul/Aug 1954

Abstract : Results of experiments on dogs revealed that increase in stability of heart to atropine in puppies as they grow older is connected with intensification of tonic excitation of the center of vagus innervation of the heart. The degree of tonic excitation of the center of vagus innervation of the heart depends on the functional condition of the peripheral endings of this vagus nerve. It evidently may also influence the quantity of acetylcholine secreted and stability to atropine. One graph. Four Soviet and two non-Soviet references.

Institution : Laboratory of Growth Physiology, Pediatrics Institute, Academy of Medical Sciences USSR, Moscow

Submitted : January 16, 1952

ARSHAVSKIY, I.A.; ROZANOVA, V.D.(Moskva)

Mechanism of phasic cardiac reactions in dogs in dysenterial intoxication in various age groups. Arkh. pat. 17 no.4:83-84
O-D '55. (MLRA 9:2)

1. Iz laboratorii vozrastnoy fiziologii i patologii (zav.-prof.
I.A. Arshavskiy) Instituta obshchey i eksperimental'noy patologii
AMN SSSR (dir.-akad. A.D. Speranskiy)

(SHIGELIA DYSENTERIAM,

antigen, eff. on heart)

(ANTIGENS AND ANTIBODIES,

Shigella dysenteriae antigen, eff. on heart)

(HEART, physiology,

eff. of Shigella dysenteriae antigen)

ROZANOVA, V.D.

Mechanism of the action of digitalis on the hearts of animals of various ages. Biul.eksp.biol. i med. 42 no.10:39-43 0 '56.
(MLRA 9:12)

1. Iz laboratori vozrastnoy fiziologii i patologii (zav. - prof. I.A.Arshavskiy) Instituta obshchey i eksperimental'noy patologii (dir. akademik A.D.Speranskiy) AMN SSSR, Moskva)

(HEART, effect of drugs on,
digitalis, age factor in animals (Rus))

(DIGITALIS, effects,
on heart, age factor (Rus))

(AGING, effects,
on heart response to digitalis in animals (Rus))

KOLHNOVA, V. D.

"Comparative Pharmacological Investigation of Hymaline Sulfate" by V. D. Rozanova, Division of Pharmacology (head, Prof A. D. Turova), All-Union Institute of Medicinal and Essential Oil-Bearing Plants, Farmakologiya i Toksikologiya, Vol 20, No 1, Jan/Feb 57, pp 52-56

Experiments were conducted on white rats, rabbits and cats in order to determine the pharmacological action and toxicity of hymaline ["gimalin"], an alkaloid isolated from Scopolia himalayan.

Hymaline, $C_{17}H_{23}O_3N$, is soluble in chloroform, alcohol, and benzene; dissolves with difficulty in ether, and is insoluble in water and petroleum ether. It has a melting point of 108-109 degrees, and readily forms sulfate, picrate, and chlorate salts. Hymaline sulfate readily dissolves in water. The results of the experiments were as follows:

SUM.1345

ROZANOVA, V.D.

(1) Hymaline was found to be an atropinelike substance with intensity 1.5-2 times that of the parasympatholytic action of atropine on the orbicular muscle of the eye, 3 times as great on cardiac cholinoreceptors, and 10 times as great on the cholinoreceptors of the gastro-intestinal tract; (2) although more intense in its action, it was found to be only slightly more toxic than atropine; (3) because of the greater intensity of its action, doses smaller than those of atropine were used in clinical experiments (0.0002 to 0.0004 milligram per kilogram of body weight).

Clinical experiments conducted at the Sanitary-Hygiene Faculty of the Therapeutic Clinic of the First Moscow Order of Lenin Medical Institute indicated that hymaline is particularly effective in the therapy of ulcers, chronic colitis, nephritic colic, and spastic conditions of the organs in the peritoneal area. Hymaline, unlike atropine, produces no side effects. (U)

S4M.1345

ROZANOVA, V.D.

Electroencephalographic analysis of peculiarities of aminazine effects in dogs of various ages. Biul.eksp.biol. i med. 47 no.6:62-67 Je '59. (MIRA 12:8)

1. Iz laboratorii vozrastnoy fiziologii i patologii (zav. - prof.I.A.Arshavskiy) Instituta normal'noy i patologicheskoy fiziologii (dir. - deystvitel'nyy chlen AMN SSSR V.N.Chernigovskiy) AMN SSSR, Moskva. Predstavlena deystvitel'nym chlenom AMN SSSR V.N.Chernigovskim.

(AGING, eff.

on EEG reactions to chlorpromazine in dogs
(Rus))

(CHLORPROMAZINE, eff.

on EEG, age factor in dogs (Rus))

(AGING, eff.

on EEG reactions to chlorpromazine in dogs
(Rus))

ROZANOVA, V.D.

Analysis of the role and nature of cholinergic and adrenergic substances of the reticular formation in dogs of various ages.
Trudy Inst.norm.i pat.fiziol. AMN SSSR 7:75-76 '64.

(MIRA 18:6)

1. Laboratoriya vozrastnoy fiziologii i patologii (zav.- prof.
I.A.Arshavskiy) Instituta normal'noy i patologicheskoy fiziologii
AMN SSSR.

ADZHIMOLAYEV, T.A.; ROZANOVA, V.D.

Mechanism of the development of inhibition (true pessimum) of
the neuromuscular apparatus in ontogenesis. Nerv. sist. no.4:
33-35 '63 (MIRA 18:1)

1. Institut normal'noy i patologicheskoy fiziologii AMN SSSR,
Moskva.

ROZANOVA, V.D.

Significance of the anabolic phase of reaction in the maintenance
of homeostasis in staphylococcal intoxication in dogs of various
age. Biul. eksp. biol. i med. 57 no.6:37-42 Je '64.

(MIRA 18:4)

I. Laboratoriya vozrastnoy fiziologii i patologii (zav. - prof.
I.A.Arshavskiy) Instituta normal'noy i patologicheskoy fiziologii
(dir. - deystvitel'nyy chlen AMN SSSR prof. V.V.Parin) AMN SSSR,
Moskva.

YENIKEYEVA, S.I.; ROZANOVA, V.D.

Mechanism of the development of coordinated (reciprocal) inhibition in ontogeny (in correlation with the analysis of some characteristics of the action of the tetanus toxin and strychnine at different ages). Trudy Inst. norm. i pat. fiziolog. AMN SSSR 6; 54-56 '62 (MIRA 17:1)

1. Laboratoriya vozrastnoy fiziologii i patologii (zav. - prof. I.A. Arshavskiy) Instituta normal'noy i patologicheskoy fiziologii AMN SSSR.

NEYMAN, M.B.; MEDZHIDOV, A.A.; ROZANTSEV, E.G.; SKRIPKO, L.A.

New reaction for forming stable Wurster salts.
Dokl. AN SSSR 154 no.2:387-390 Ja'64. (MIRA 17:2)

1. Institut khimicheskoy fiziki AN SSSR. Predstavлено
академиком А.А. Баландиным.

ARSHAVSKIY, I.A.; ROZANOVA, V.D.

Mechanism of phasic reactions of the heart in dogs during dysentery intoxication at different age periods. Trudy Inst. norm. i pat. fiziol. AMN SSSR no.1:69-81 '58 (MIRA 16:12)

1. Iz laboratorii vozrastnoy patofiziologii (zav. - prof. I.A. Arshavskiy) otdela ebschey i eksperimental'noy patologii (zav. - akademik A.D. Speranskiy) Instituta normal'noy i patologicheskoy fiziologii AMN SSSR.

ROZANOVA, V.D.

Mechanism of the development of anesthetic inhibition in ontogeny. Trudy Inst. norm. i pat. fiziolog. AMN SSSR 6:50-53 '62
(MIRA 17:I)

1. Laboratoriya vozrastnoy fiziologii i patologii (zav. - prof. I.A.Arshavskiy) Instituta normal'noy i patologicheskoy fiziologii AMN SSSR.

ROZANOVA, V.D.

Sensitivity, resistance, and tolerance of rats of various ages to tetanus toxin. Biul. eksp.biol.i med. 54. no.12: 36-42 D'62. (MIRA 16:6)

1. Iz laboratorii vozrastnoy fiziologii i patologii (zav. - prof. I.A.Arshavskiy) Instituta normal'noy i patologicheskoy fiziologii (dir. - deystvitel'nyy chlen AMN SSSR prof. V.V.Parin) AMN SSSR, Moskva. Predstavlena deystvitel'nym chlenom AMN SSSR V.V.Parinym. (TOXINS AND ANTITOXINS) (TETANUS) (AGING)

ROZANOVA, V.D., (Moskva)

Resistance and tolerance in relation to certain pharmacological substances at different age periods. Usp. sovr. biol. 49 no.1: 86-103 Ja-F '60. (MIRA 14:5)

(DRUGS—PHYSIOLOGICAL EFFECT) (AGE).

ROZANOVA, V.D.

Formation of gastric ulcers in pregnant and nonpregnant rats in
the light of data on the dominant of gestation. Biul.eksp. biol.
i med. 51 no.1:40-43 Ja '61. (MIRA 14:5)

1. Iz laboratorii vozrastnoy fiziologii patologii (zav. - prof.
I.A.Arshavskiy) Instituta normal'noy i patologicheskoy fiziologii
(dir. - deystvitel'nyy chlen AMN SSSR V.V.Parin) AMN SSSR, Moskva.
Predstavlena deystvitel'nym chlenom AMN SSSR V.V.Parinym.
(PEPTIC-ULCER) (PREGNANCY, COMPLICATIONS OF)
(CAFFEINE—PHYSIOLOGICAL EFFECT)

ROZANOVA, V.D.

Nervous and reflex mechanism of the specific action of digitalis
on the cardiovascular system at various ages. Trudy Vses. ob-va
fiziol., biokhim. i farm. 4:160-166 '58. (MIRA 14:2)

1. Laboratoriya vozrastnoy fiziologii Instituta obshchey i
eksperimental'noy patologii AMN SSSR (zav. laboratoriye prof.
A.I. Arshavskiy).
(DIGITALIS) (CARDIOVASCULAR SYSTEM) (REFLEXES)

ROZANOVA, V.D.

Criteria of sensitivity, resistance, and tolerance in the effect
of certain pharmacological substances during various stages of
development. Biul. eksp. biol. i med. 49 no. 5:87-93 My '60.
(MIRA 13:12.)

1. Iz laboratorii vozrastnoy fiziologii i patologii (zav. - prof.
I.A. Arshavskiy) Instituta normal'noy i patologicheskoy fiziologii
(dir. - deystvitel'nyy chlen AMN SSSR V.N. Chernigovskiy) AMN
SSSR, Moskva. Predstavlena deystvitel'nym chlenom AMN SSSR V.N.
Chernigovskim.

(AGING) (PHARMACOLOGY)

ROZANOV, V.I., Cand Tech Sci -- (disc) "Study of the heat ^{resistance} stability of annealed and tempered glasses." M.S., 1958, 14 pp (Min of Higher Education USSR. Mos Order of Lenin Chem Tech Inst im D.I. Mendeleyev)
150 copies (KL, 27-53, 111)

SOV/179-59-2-25/40

AUTHORS: Bartenev, G. M., Rozanova, V. I. (Moscow)

TITLE: Thermal Endurance and Strength of Glass (Termostoykost' i prochnost' stekla)

PERIODICAL: Izvestiya Akademii nauk SSSR OTN, Mekhanika i mashino-stroyeniye, 1959, Nr 2, pp 159-162 (USSR)

ABSTRACT: The paper is a continuation of previous work (Refs 3-6), in which the maximum thermal stress developed on two-sided cooling of a glass plate was calculated. This solution shows that the thermal stress attains a maximum value with time (S_m^{-1}) and the thermal endurance is defined as

$$\sigma = \frac{P(1 - \mu)}{\beta E} \frac{1}{haS_m} \quad (1)$$

where P , μ , β , E , h and a are respectively the ultimate strength, the Poisson's ratio, the coefficient of thermal expansion, the Young's modulus, the coefficient of heat emission and the half-thickness of the glass. The thermal endurance of a number of specimens was measured on cooling in air and in water. The effect of thickness, of hardening and of annealing was investigated, and the results are presented in the form of graphs. The character of the

Card 1/2

SOV/179-59-2-25/40

Thermal Endurance and Strength of Glass

rupture is described both for annealed and for hardened glass, and photographs are reproduced showing various hardened glasses after fracture. There are 7 figures and 7 references, of which 5 are Soviet and 2 English.

ASSOCIATION: Institut stekla (Glass Institute)

SUBMITTED: March 29, 1959 .

Card 2/2

BARTENEV, G.M. (Moskva); ROZANOVA, V.I. (Moskva)

Thermal endurance and strength of glass. Izv.AN SSSR.Otd.
tekhn.nauk.Mekh. i mashinostr. no.2:159-162 Mr-Ap '59.
(MIRA 12:5)

1. Institut stekla AN SSSR.
(Glass research)

ROZANOVA, V.I.

Methods and some results of heat resistance testing of glass. Trudy
VNIIStekla no.37:59-66 '57. (MIRA 11:1)
(Glass--Testing)

MOZANOV V.I.

21 15
Thermal endurance of glass on cooling. G. M. Bartenev
and V. I. Rozanova. *Izvest. Akad. Nauk S.S.R., Otdel.*
Tekh. Nauk 1957, No. 5, 62-9.—The thermal endurance of
glass was studied by gradually raising its temp. in 5° steps
to the required temp., keeping it at that temp. for 30 min.,
and then suddenly chilling it, either by plunging into water
at 17°, or directing an air blast at 20° on it. The temp. to
which the glass was heated when it cracked during the chilling
was taken as its endurance temp., and was an av. of 30
degrs. The endurance is a more complex characteristic
than its strength and includes besides the strength a no. of
other factors, such as the size of the sample, the rate of cool-
ing, etc. A formula for calculating the endurance of flat glass
samples was derived, and was similar to the formula derived
by Tabata and Moriya (*C.A.* 28, 505¹, 2141¹) for round rods.
W. M. Sternberg

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RW
MT

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ROZANOVA, V. I.

Journal of Applied Chemistry
June 1954
Industrial Inorganic Chemistry

Simple optical methods of measuring the elasticity constant of glass. G. M. Burtsev and V. I. Rozanova (Glass & Ceramics, Moscow, 1953, 10, No. 6, 13).—The optical constant was measured by a simple method of bending the glass specimen. Concentration of stresses around the point of pressure, where the load rests, does not give a true picture of the main load distribution. Consequently tensile stresses on the opposite surface (against the point of load) are calculated. The optical constant depends neither on the stress value nor on whether the stress is positive or negative. It is concluded that the optical constant can be determined on non-etched specimens under a convenient stress and that data obtained for low stresses can be used for calculating the strength of tempered glasses.

BRIT. CERAM. ABSTR. (R.B.C.).

MF
N-10-5

Rozanova, V.I.

✓ Simple methods of measuring optical constant of elasticity of glass. G.M. BARTENY AND V.I. ROZANOVA. *Sieklo i Keram.*, 10 (10) 13-15 (1953).—Measurements were made by bending a M/T glass rod and viewing with a polarimeter. For loads greater than breaking, specimens were treated with HF. Calculations were made from constant $B = 1/3 \times d^3/aF \times R \times 10^{-1}$ where d is thickness (cm.), a is distance from point of application of load to the support (two concentrated loads were used), and R is full difference of the movement recorded by the compensator ($m\mu$). Cf. *Ceram. Abstr.*, 1955, April, p. 67h. B.Z.K.

1/18 R2

AUTHORS: Bartenev, G.M. and Rozanova, V.I. (Moscow)

24-5-7/25

TITLE: Heat resistance of glass during cooling.
(Termostoykost' stekla pri okhlazhdenii).

PERIODICAL: "Izvestiya Akademii Nauk, Otdeleniye Tekhnicheskikh Nauk"
(Bulletin of the Ac.Sc., Technical Sciences Section),
1957, No.5, pp.62-69 (U.S.S.R.)

ABSTRACT: The authors investigated the heat resistance of sheet glass which for a certain time was heated at a constant temperature in a furnace and then rapidly cooled in water or by an air jet. The time of movement of the glass from the furnace into the water bath or to the air jet was 1 to 2 sec. The tests consisted of increasing gradually the furnace temperature by steps of 5°C and the temperature at which the glass broke up was taken as a measure of the heat resistance. It was found that the glass can withstand higher temperature gradients in the case of unilateral cooling than if cooling is applied from both sides. The test results for water and air for a 6 mm thick glass are given in Fig.1. Fig.2 shows the influence of the heating time on the results for a glass of equal thickness. Figs. 3 and 4 show the pattern of fracture of the investigated glass, the graph, Fig.5, shows the influence of the submersion speed on the heat resistance

Card 1/2

Rozanova V.I.

✓ Formula for calculating the extent of hardening of flat glass as a function of the thickness. G. M. BARTHEV AND V. I. ROZANOVA. *Siekly i Keram.*, 9 [10] 6-7 (1952).—The extent of hardening, Δ , in μ/cm . (d is one-half the glass thickness) can be calculated from the following approximate formula: $\Delta = A \frac{hd}{6+hd}$ where d is thickness of glass (cm.), h is relative coefficient of heat loss during hardening, and $A = 1/sB \times 10^3 \frac{\beta E}{1-\sigma} T_g$, where B is optical constant of stresses ($\text{cm.}^2/\text{kg.}$); β is coefficient of linear expansion; E is Young's modulus; σ is Poisson's coefficient; and T_g is temperature of vitrification calculated from the temperature of the hardening medium. This formula was checked against experimental data obtained with glass of SiO_2 70.9, Na_2O 16.1, K_2O 0.6, CaO 8.1, MgO 2.0, Al_2O_3 0.8, SO_3 0.6, and Fe_2O_3 0.1%. Three conditions of cooling were observed: (1) natural convection ($p = 0$), (2) air stream (excess pressures of $p = 50$ mm. H_2O and 60 mm. Hg), and (3) distance from nozzle to glass, 50 cm. Calculated and experimental results were alike. For most commercial glasses, $A \approx 5 \times 10^3$ and the above formula becomes $\Delta \approx 5 \times 10^3 \frac{cd}{6+cd}$, where h is replaced by the empirical constant c having the same dimension, and Δ is given in μ/cm . and d in cm. If Δ is expressed in $N/\text{cm.}$, then the formula becomes $\Delta \approx \frac{9cd}{6+cd}$, where c is a constant in $1/\text{cm.}$ This formula is recommended in technology and industry. The constant c is calculated from $c = 0A/(9 - \Delta)d$. H.Z.K.

3
300
LM

109. Simple optical methods of measuring the elasticity constant of glass.—G. M. BARTENYEV and V. I. ROZANOVA (*Glass & Ceramics*, Moscow, 10, No. 6, 13, 1953). There are no direct methods of measuring internal stresses in glass. When tempered or annealed glasses are studied by means of a polarimeter, it is the optical difference that is measured and not the stresses; for subsequent calculations of the stresses the optical stress constant for the given glass is required. In this work the optical constant was measured by a simple method of bending the glass specimen. Concentration of stresses around the point of pressure where the load rests, does not give the true picture of the main load distribution. Consequently tensile stresses on the opposite surface (against the point of load) were calculated. It was found that the optical constant depends neither on the stress value nor on whether the stress is positive or negative. It is concluded that the optical constant can be determined on non-etched specimens under a convenient stress and that the data so far obtained for low stresses can be used for calculating the strength of tempered glasses. (4 figs.)

*AP
Jew*

Rozanova, V. I.

USSR

Formula for calculating the extent of hardening of glass as a function of thickness. G. M. Bartenev and V. I. Rozanova. *Steklo i Keram.* 9, No. 10, 6-7(1952). The extent of hardening Δ in μ/cm . (a is $\frac{1}{4}$ of glass thickness) can be calcd. from the following approx. formula: $\Delta = A(hd/6 + cd)$ where d is thickness of glass in cm., h is relative coeff. of heat loss during hardening, and $A = \frac{1}{4}B \times 10^4(\beta E/1 - \sigma T_g)$, where B is optical const. of stresses in sq. cm./kg., β is coeff. of linear expansion, E is Young's modulus, σ is Poisson's coeff., and T_g is temp. of vitrification calcd. from the temp. of the hardening medium. This formula was checked against exptl. data obtained with glass of SiO_2 70.9, Na_2O 16.1, K_2O 0.8, CaO 8.1, MgO 2.9, Al_2O_3 0.8, SO_3 0.6, and Fe_2O_3 0.1%. Three conditions of cooling were observed: (1) natural convection ($\rho = 0$), (2) air stream (excess pressures of $\rho = 50$ mm. H_2O and 50 mm. Hg), and (3) distance from nozzle to glass 50 cm. Calcd. and exptl. results were alike. For most com. glasses, $A \approx 5 \times 10^4$ and above, formula becomes $\Delta \approx 5 \times 10^4(cd/6 + cd)$, where h is replaced by empirical const. c having the same dimension and d is in cm. and Δ is in μ/cm . If Δ is expressed in N/cm , then the formula becomes $\Delta \approx 9cd/6 + cd$, where c is a const. in $1/\text{cm}$. This formula is recommended for use in industry and technology. The const. c is calcd. from $c = \frac{6.7}{d} - \Delta M$.

(1)

62

ROZANOVA, V. I.

Journal of Applied Chemistry
Vol. 4 Feb. 1954
Industrial Inorganic Chemistry

Correlation between the degree of hardening and the thickness of flat glass. G. M. Bartenev and V. I. Rozanova (*Steklo i Keram.*, 1952, 9, 6; Summary, *Glass Ind.*, 1953, 34, 545-546).—Sheets 30 × 50 or 85 × 130 mm. and 2-26 mm. thick were cooled in still air or in air blasts of known pressure. The degree of hardening, Δ , was measured in polarised light as the path difference for 1 cm. length. The results confirmed the theoretically derived relationship: $\Delta = B\alpha ETghd/(6 + \alpha d)$, where B is the optical stress constant, α the thermal expansion, E the modulus of elasticity, T the difference between the cooling air and the softening temp., α Poisson's ratio, h the coeff. of heat loss, and d the thickness of the glass. For the glass SiO_2 70.9, Na_2O 16.1, K_2O 0.6, CaO 8.1, MgO 2.9, Al_2O_3 0.8, SO_3 0.5, and Fe_2O_3 0.1%, B was 2.58×10^{-7} sq. cm. per kg. and $\alpha E/(1 - \alpha)$ was 7.80 kg. per sq. cm. per $^{\circ}\text{C}$. The simplified relationship, $\Delta = 5cd/(6 + cd)$, where c is an empirical constant related to h , is applicable to ordinary commercial glasses. Values of c are given for air-blast pressures of 0-800 mm. (water-gauge).

J. A. SUGDEN

11-105

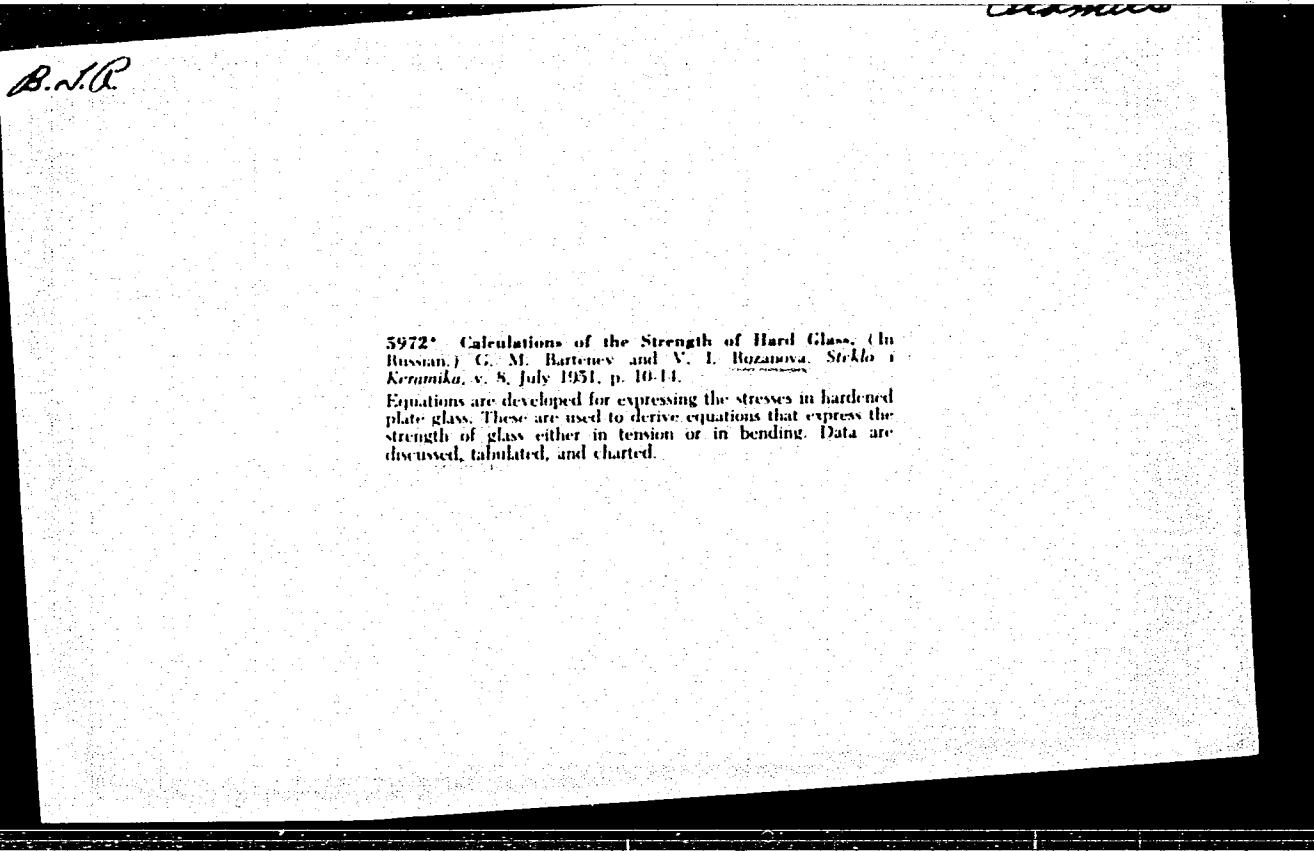
1. BARTENEV, G. M. and ROZANOVA, V. I.
2. USSR (600)
4. Plate Glass
7. Formula for calculating the extent of hardening of plate glass on the basis of thickness, Stek.i ker 9 No. 10, 1952.
9. Monthly List of Russian Accessions, Library of Congress, February 1953. Unclassified.

BARTENEV, G.M.; ROZANOVA, V.I.

Simple methods for measuring the optical constant of glass elasticity.
Stek. i ker. 10 no.6:13-15 Je '53. (MLRA 6:5)
(Glass)

4-52

Calculations of the strength of tempered glass. G. M. BARTENYEV AND V. I. ROZANOVA. *Nikolskii Keram.*, 8 [7] 10-11 (1951).—Resistance P (in kg/cm^2) of tempered sheet against bending and stretch is expressed by $P = P_0 - \frac{x}{B \cdot 10^3} \cdot \Delta$ where P_0 is resistance of untempered glass (in kg/cm^2), Δ is extent of tempering in m^2/cm^2 , B is optical stress constant (in cm^2/kg), and x is dimensionless coefficient. This equation is also applicable to rods, cylinders, etc., the only difference being that x has different values. The values are given of x as applicable to a sheet. This equation is valid if the shape is tempered, with ends being insulated. If the ends of the shape are open, the equation becomes $P = K(P_0 - \frac{x}{B \cdot 10^3} \cdot \Delta)$, where K is a correction coefficient. Both equations are applicable up to 5 N/cm^2 of tempering where $N = 540 \text{ m}^2$.



5972* Calculations of the Strength of Hard Glass. (In Russian.) G. M. Bartenev and V. I. Bozanova. *Steklo i Keramika*, v. 8, July 1951, p. 10-14.

Equations are developed for expressing the stresses in hardened plate glass. These are used to derive equations that express the strength of glass either in tension or in bending. Data are discussed, tabulated, and charted.

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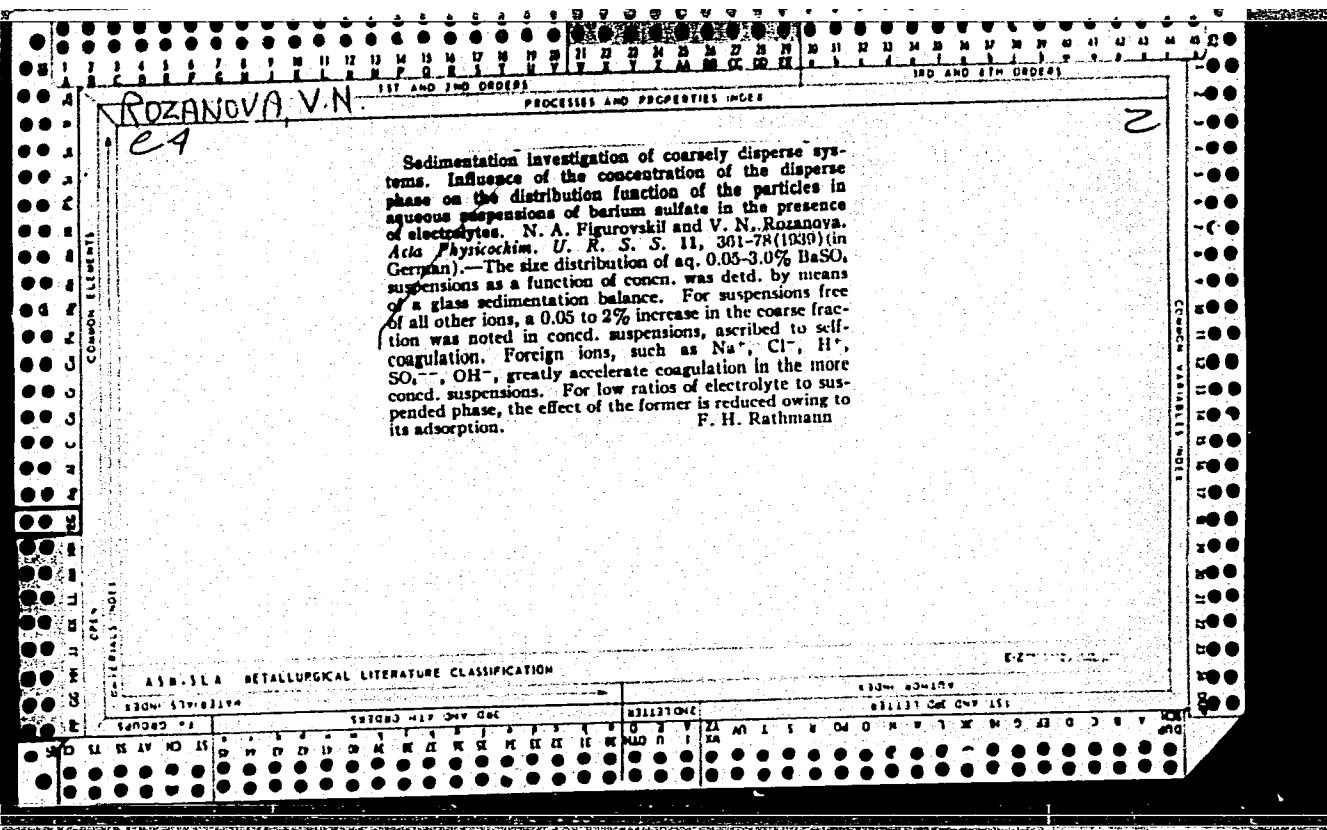
Calculations of the strength of tempered glass. G. M. Bartenev and V. I. Roganova. *Steklo i Keram.* 8, No. 7, 10-14(1951).—Resistance P (in kg./sq. cm.) of tempered sheet against bending and stretch is expressed by $P = P_0 - (x/B \times 10^7)\Delta$, where P_0 is resistance of untempered glass in kg./sq. cm., Δ is extent of tempering in ms/cm., B is optical stress const. in sq. cm./kg., and x is a dimensionless coeff. This equation is also applicable to rods, cylinders, etc., the only difference being that x has different values. Values of x are given for a glass sheet. This equation is valid if the shape is tempered, with ends being insulated. If the ends of the shape are open, the equation becomes $P = K(P_0 - (x/B \times 10^7)\Delta)$, where K is a correction coeff. Both equations are applicable for up to $\delta N/cm.$ of tempering, where $N = 540$ ms.

1952

BCS

*Ceramic Products
Glass*

309. Calculations of the strength of tempered glass.—G. M. BARTENEV and V. I. ROZANOV (*Stek. Keram.*, 8, No. 7, 10, 1951). Starting from the principle that the strength of glass is determined by the strength of its surface and using formulae of the theory of glass tempering, a formula is obtained for the calculation of the transverse and tensile strengths of flat tempered glass. In this formula the coeff. X , which is equal to the ratio of the absolute value of the compression on the surface to the expansion inside the glass, depends on the degree of tempering. For flat glass the calculated values for X are tabulated. The formula recommended can also be used for specimens of different shapes, provided that the values for X are then found from different tempering formulae. A comparison of the calculations with exptl. results has shown that the formula recommended is applicable to flat specimens tempered with edges covered with thermal insulation (a 6-mm. layer of asbestos), whereas the strength of specimens tempered with uninsulated edges was higher than the calculated value. A correction coeff. is recommended for specimens tempered with the edges uninsulated. The limits within which certain formulae can be applied are discussed. (2 figs., 2 tables.)



ROZANOVA, V. N.

"Metric-Sediment Investigations of Large Dispersion Systems: Influence of the Dispersal Phase on the Distribution Function of Small Parts of Hydrous Suspensions of Barium-Sulphate in the Presence of Electrolytes,"
Acta Phys., 11, No. 3, 1939.

ROZANOVA, V. N.; FIGUROVSKIY, N. A.

Physical Chemistry of Dispersion Systems Department, Colloids and Electro-
chemical Institute, Academy of Sciences USSR, Moscow, (-1939-).

"Sedimentometric Investigations of Coarse Dispersion (Grubodispersnyy) Systems".

Zhur. Fiz. Khim., Vol. 14, No. 1, 1940

ROZANOV, E. P.

1728. Materialy K Razrabotke Gigienicheskikh Normativov Pit'evogo Regima Rabochikh V Ustroevyakh Zavodakh Olastey Sovetskogo Soyuza. N. 1954.
128. 1954. (Akad. Ned. Nauk SSSR.) 100 eks. B. ts.--(54-52136)

SO: Knizhnaya Letopis' Vol. 1, 1955

Rozanova Ye.F.

DAVYDOV, V.G.; BIRZHEVAYA, M.G.; IVANOVA, N.I.; KEPPOVA, Z.V.; NAUMOVA, A.Y.;
ROZANOVA, Ye.F.; SADKOVSKAYA, N.I.

Hygienic measures for preventing the overheating of the body while
working under hot climatic conditions. Gig.i san. no.5:18-23 My '54.
(MLRA 7:5)

1. Iz Instituta gigiyeny truda i professional'nykh zabolеваний
Akademii meditsinskikh nauk SSSR. (Heat—Physiological effect)
(Industrial hygiene)

ROZANOVA, Ye. F.

"Data on the Development of Hygienic Norms for the Drinking Schedule of Laborers Under the Conditions Prevailing in the Torrid Regions of the Soviet Union." Cand Med Sci, Acad Med Sci USSR, 29 Dec 54. (VM, 14 Dec 54)

Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions (12)

SO: SUM No. 556, 24 Jun 55

ROZANOVA, Ye. F.

"Data on the Development of Hygienic Standards for a Nutritional Regime for Workers in the Hot Oblasts of the Soviet Union." Cand Med Sci, Acad Med Sci USSR, Moscow, 1954. (KL, No 1, Jan 55)

Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions (12)
SO: Sum. No. 556, 24 Jun 55

ROZANOVA, Ya.V.

Distribution of micro-organisms in the Borislav oococrite
deposits. Mikrobiologija 33 no.2:306-313 Mr-Apr '64.
(MEHA 17:12)

1. Institut mikrobiologii AN SSSR.

ROZANOWSKI, Jerzy, mgr inż.

Repairs of typ fied ships produced in series. Bud. okretowe
Warszawa 10 no.3:98-100 Mr '65.

1. Department of Ship Repairs of the Marine Institute, Giansk.

ROZANOVA, Ye.P.; SHTURM, L.D.

Species of micro-organisms isolated from the Borislav ozocerite deposits and their relation to the components of petroleum and ozocerite. Mikrobiologija 33 no.1:126-133 Ja-F '64.

(MIRA 17:9)

1. Institut mikrobiologii AN SSSR.

ROZANOVA, Ye.P., NOVOZHILOVA, M.I.

Quantitative distribution and specific composition of yeasts in
Rybinsk Reservoir [with summary in English]. Mikrobiologija
27 no.3:371-376 My-Je '58 (MIRA 11:9)

1. Institut biologii vodokhranilishch AN SSSR "Borok."
(WATER SUPPLY, microbiology,
yeasts in reservoirs (Rus))
(YEASTS,
in water reservoirs (Rus))

SOROKIN, Yu.I.; ROZANOVA, Ye.P.; SOKOLOVA, G.A.

Studying primary production in Gorkiy Reservoir by the use of
 C^{14} . Trudy Gidrobiol. ob-va 9:351-359 '59. (MIRA 12:9)

1. Institut biologii vodeokhranilishch AN SSSR.
(Gorkiy Reservoir--Photosynthesis)

ROZANOVA, Ye.P.; SHTURM, L.D.

Studying amino acids released by micro-organisms into the petroleum-bearing culture medium with reference to the genesis of ozoceritelike bitumens. Mikrobiologija 29 no.5:710-714 S-0 '60. (MIRA 13:11)

1. Institut mikrobiologii AN SSSR.
(MINUSINSK-BASIN--BITUMEN--MICROBIOLOGY)
(PETROLEUM--MICROBIOLOGY) (AMINO ACIDS)

SHTURM, L.D.; ROZANOVA, Ye.P.

Study of the micro-organisms of the Minusinsk Basin in connection
with the genesis of ozocerite bitumens. Mikrobiologiya 30 no.1:
122-129 Ja-F '61.
(MINUSINSK BASIN--MICRO-ORGANISMS) (BITUMEN)

ROZANOVA, Ye.P.; SHTURM, L.D.

Microflora of the Shor-Su ozocerite deposits. Mikrobiologija
34 no.5:888-894 S-0 '65. (MIRA 18:10)

1. Institut mikrobiologii AM SSSR.

ROZANOVA, Ye.A.

Microbiological study of oils and formation waters of the
Borislav fold as related to the explanation of the role
of micro-organisms in the genesis of ozocerite. Mikrobiologija
33 no.4:692-698 Jl-Ag '64. (MIRA 18:3)

i. Institut mikrobiologii AM SSSR.

SHTURM, L.D.; ROZANOVA, Ye.P.

Study of yeasts of the *Candida* genus, developing on hydrocarbons,
isolated from ozocerite beds. *Mikrobiologija* 32 no.6:1013-1019
N-D '63 (MIRA 18:1)

1. Institut mikrobiologii AN SSSR.

KOVNER, A.A., dots., ROZANOVA, Ye.K., TOPEL'BERG, M.S.

Analysis of the operation of a stomatological polyclinic. Stomatologija
37 no.6:53-56 N-D '58 (MIRA 11:12)

1. Iz kafedry organizatsii zdravookhraneniya Moskovskogo meditsinskogo
stomatologicheskogo instituta (zav. kafedroy i dir. instituta - dots.
G.N. Beletskiy).
(STOMATOLOGY)

CHERNOV, O.I., inah.; ROZANTSEV, Ye.S., inzh.; PUZYREV, V.N., inzh.

Sudden coal and gas outbursts in Karaganda mines. Bezop.truda v
prom. 5 no.4:4-6 Ap '61. (MIRA 14:3)

1. Vostochnyy nauchno-issledovatel'skiy institut po bezopasnosti
truda v gornoj promyshlennosti.

(Karaganda Basin—Coal mines and mining—Accidents)

CHERNOV, O.I.; ROZANTSEV, Ye.S.

First sudden gas and coal outbursts during stoping operations
in the Kuznetsk Basin. Ugol' 36 no.4:44 Ap '61. (MIRA 14:5)

1. Vostochnyy nauchno-issledovatel'skiy institut po bezopasnosti
rabot v gornoj promyshlennosti.
(Kuznetsk Basin—Mine gases)

ROZANOVA, Yu. M.

Isoquinoline compounds. III. Synthesis of 2-methyl-1-(3,4-dimethoxybenzyl)-5,6-dimethoxy-1,2,3,4-tetrahydroisoquinoline. E. S. Livshits, I. S. Painova, G. I. Bazilevskaya, E. I. Genkin, N. A. Preobrazhenskii, Yu. M. Rozanova, and Z. A. Barenova (M. V. Lomonosov State Univ., Moscow). Zhur. Obshchei Khim. (J. Gen. Chem.) 21, 1354-60 (1951); cf. C.A. 42, 2606g; 43, 2212i. Guaiacol allyl ether (74% from $\text{CH}_2:\text{CHCH}_2\text{Cl}$) and guaiacol, b_{13} 111-13°, d_{20}^{20} 1.0592, n_D^{20} 1.5362, heated 3 hrs. to 230° gave 65% 2-hydroxy-3-methoxy-1-allylbenzene, b_{15} 124-5°, d_{20}^{20} 1.0904, n_D^{20} 1.5411, which, heated with KOH 5 hrs. to 170°, gave 52% 1-(2-hydroxy-3-methoxyphenyl)propene, b_9 125-8°, d_{20}^{20} 1.0372, n_D^{20} 1.5535, m. 66°. This with Me_2SO_4 and aq. NaOH gave the Me ether, b_{11} 128°, d_{20}^{20} 1.0372, n_D^{20} 1.5535. This (25 g.) in 480 ml. H_2O and 58 g. $\text{K}_2\text{Cr}_2\text{O}_7$ treated with 45 ml. concd. H_2SO_4 at 38-40° gave 70% 2,3-(MeO)₂C₆H₃CHO, b_{11} 133-5°, m. 52-3°, which with $\text{CH}_2(\text{CO}_2\text{H})_2$ gave 93.6% 2,3-(MeO)₂C₆H₃CH:CHCO₂H, m. 177°, yielding with Na-Hg 85% of the propionic acid, m. 68°, which with MeOH-H₂SO₄ gave 90% Me ester, b_9 154-5°, d_{20}^{20} 1.127, n_D^{20} 1.5130. This with satd. NH_4OH gave 86% amide, m. 99-100°, yielding with Fr-KOH 72.5% 2,3-(MeO)₂C₆H₃CH₂-CH₂NH₂ (I), b_9 134-5°. Vanillin and Me_2SO_4 gave 90% 3,4-(MeO)₂C₆H₃CHO, which with 40% formalin in H_2O -EtOH in the presence of KOH yielded 96% 3,4-(MeO)₂-C₆H₂CH₂OH, b_{10} 159-60°; this and SCCl_2 gave 90% of the corresponding chloride, m. 51°, which with KCN yielded 68% cyanide, b_9 , 168-70°, hydrolyzed to the acid(87%), m. 98-9° (Et ester 85%), b_5 159-60°. The ester (3.75 g.) and 3 g. I with a few drops of pyridine heated 3 hrs. at 180° gave 62.3% N-(2,3-dimethoxyphenethyl) (3,4-dimethoxyphenyl)acetamide, m. 89° (from petr. ether). This heated with PCl_3 2 hrs. at 100° gave 65% 3,4-dimethoxyphenyl 5,6-dimethoxy-3,4-dihydro-1-isouinolyl ketone, m. 119.5° (from EtOH), which with MeI gave the methiodide, m. 178-9° (from EtOH). The latter (1 g.) with Zn dust (cf. C.A. 42, 2606g) gave 69.6% 2-methyl-1-(3,4-dimethoxybenzyl)-5,6-dimethoxy-1,2,3,4-tetrahydroisoquinoline, m. 76-7° (from ligroine); HCl salt, m. 139-41°. The compd. is

a model substance in the synthetic studies on morphine.

IV. Synthesis of 1-(2-(3-pyridyl)ethyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline.
R. S. Livshits, E. P. Evstigneeva, M. S. Bainova, and N. A. Freobrazhenskii. Ibid. 136C-
4.-Conventional esterification gave 80% Me nicotinate, b7 89-90°, m. 38°, which with
N₂H₄.H₂O gave 98-9% hydrazide, m. 158-9°; this with EzCl at 0° gave 96.5% N-Ez deriv.,
m. 185-6°, which, heated in (CH₂OH)₂ with Na₂CO₃ 2 min. to 160° yielded 28-30%
nicotinaldehyde, b12 85-90°, condensed with CH₂(CO₂H)₂ to 3-pyridineacrylic acid,
m. 232-3° (from EtOH). This (8g.), 40 ml. AcOH, 80 ml. HI (d. 1.71), and 3 g. red
P refluxed 14-15 hrs. gave 90% 3-pyridinepropionic acid-HI, m. 163-4°, yielding
with Na₂HPO₄ the free acid, m. 157-8° (from EtOH) (Et ester (75.7% with EtCH-HCl),
b7 129-30°, n_D²⁰ 1.4983, d₂₀²⁰ 1.071; HCl salt, m. 95-6°; picrate, m. 81-2°). The
Et ester (3g.) and 3 g. 3,4-(MeO)₂C₆H₃CH₂CH₂NH₂ heated with a few drops of pyridine
3 hrs. at 180° gave 80.7% N-(3,4-dimethoxyphenethyl)-3-pyridinepropionamide, m. 103°,
which heated with POCl₃ 2.5 hrs. at 100° yielded 85% 1-(2-(3-pyridyl)ethyl)-6,7-
dimethoxy-3,4-dihydroisoquinoline, m. 88-9° (from petr. ether); HCl salt, m. 198-200°
(from EtOH). The free base (1 g.), 30 ml. H₂O, 4 g. Zn dust, and 0.1 g. CuSO₄ treated
with 3 n.l. concd. H₂SO₄ and gently heated 1 hr., then heated more strongly 2 hrs.,
gave 70% 1-(2-(3-pyridyl)ethyl)-6,7-dimethoxy-1,2,3,4-tetrahydroisoquinoline, an oil;
HCl salt, m. 241-3° (from EtOH).

G. M. Kosolapoff

NATRADZE, A.G., kand. tekhn. nauk; ARONSON, Yu.P.; ROZEN, I.F.; ROZANOVA, Yu.M.; ZOLOTNITSKIY, I.M., red.; KNAKNIN, M.T., tekhn. red.

[Protecting chemical apparatus from corrosion in pharmaceutical plants] Zashchita khimicheskoi apparatury ot korrozii v khimiko-farmatsevticheskoi promyshlennosti. Pod obshchei red. A.G. Natradze. Moskva, Gos. izd-vo med. lit-ry, 1958. 283 p.

(Drug industry) (MIRA 11:9)

(Corrosion and anticorrosives)
(Protective coatings)

ARONSON, Yu.P.; ROZANOVA, Yu.M.

Corrosion resistance of some metals and alloys in organic acids.

Khim. i med. no. 12:117-131 '59. (MIRA 13:10)

(METALS—CORROSION) (ACIDS, ORGANIC)

5.3610,5.3620

77533
SOV/80-33-1-42/49

AUTHORS: Rozanova, Yu. M., Aronson, Yu. P.

TITLE: Investigation of the Possibility of Removal of the
Adverse Effect of Iron Compounds in the Preparation
of 2-Amino-2-mercaptop-1,3,4-thiodiazole

PERIODICAL: Zhurnal prikladnoy khimii, 1960, Vol 33, Nr 1, pp
233-237 (USSR)

ABSTRACT: It is shown that on cyclization of $\text{H}_2\text{NCSNNHCNSNH}_2$
(I) in the presence of iron, the yield of 2-amino-
5-mercaptop-1,3,4-thiodiazole (II) (main product)
decreases from 39-40 to 20-24%. In the presence of
metallic granular tin (or powder), 20% HCl (chemically
pure), and iron, the yield of (II) increases from
20-24 to 65-67%, and in the absence of iron the yield
increases from 39-40 to 65-67%. The use of metallic
tin in the above reaction makes possible the use of
technical instead of cp HCl. The yield in the above

Card 1/5

Investigation of the Possibility of Removal
of the Adverse Effect of Iron Compounds in
the Preparation of 2-Amino-2-mercaptopro-1,3,
4-thiodiazole

77533

SOV/80-33-1-42/49

case increases from 18-22 to 65-67%. The addition of tin in the form of powder is recommended. Tin can be easily recovered with granular Zn from the mother liquor. The use of Zn dust, SnCl_2 , granular Zn, does not give satisfactory results. The cost of Zn and Sn is 6 rubles per 1 kg. of I. On substitution of technical for pure HCl the saving will be 200 rubles per 1 kg of I. There are 2 tables; and 2 references, 1 German, 1 U.S. The U.S. reference is: Am. Pat. 2759947, October 27, 1955.

ASSOCIATION:

Ordzhonikidze All-State Scientific Research Chemical-
Pharmaceutical Institute (Vsesoyuznyy nauchno-
issledovatel'skiy khimiko-farmatsevticheskiy institut
imeni S. Ordzhonikidze)

SUBMITTED:
Card 2/5

September 13, 1958

S/081/62/000/006/048/117
B149/B108

AUTHORS: Aronson, Yu. P., Rozanova, Yu. M.

TITLE: Resistance of some metals and alloys to corrosion by organic acids

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 6, 1962, 353, abstract 6I208 (Sb. "Khimiya i meditsina" no. 12, M., Medgiz, 1959, 117 - 131)

TEXT: The corrosion resistance was tested of X18H12M3T (Kh18N12MZT) and 1X18H9T (1Kh18N9T) steels in boiling CH_3COOH of various concentrations and of Cu in boiling formic acid in air and in a neutral atmosphere. It was found that the resistance of stainless steel to CH_3COOH is increased if the acid is purified by addition of 2.5 % CrO_3 . 100 % CH_3COOH , obtained by concentrating glacial acetic acid with acetic anhydride has the highest corrosive action. Even Kh18N12MZT steel is not resistant to this acid in the vapor phase. This steel can be used with 100 % CH_3COOH in the liquid

Card 1/2

Resistance of some metals and ...

S/081/62/000/006/048/117
B149/B108

phase and under all the other conditions investigated. 1Kh18N9T steel cannot be used in the vapor phase of acid of 80 % or higher concentration. It is resistant to the liquid phase of 10, 50, 80, 90, and 98 % CH₃COOH.

The corrosion of Cu in boiling formic acid can be decreased by blowing oxygen-free nitrogen through the acid. Under these conditions Cu is resistant to 50, 76 and 80 % boiling formic acid. The rate of its corrosion is not higher than 0.1 mm per year. [Abstracter's note: Complete translation.]

Card 2/2

✓
55

S/081/62/000/021/047/069
B171/B101

AUTHORS:

Krol', B. B., Zherdeva, L. G., Rozanova, Z. I.,
Rozhdestvenskaya, A. A.

TITLE:

Effects of sulfur compounds on the stability of transformer
oils

PERIODICAL:

Referativnyy zhurnal. Khimiya, no. 21, 1962, 404, abstract
21M146 (Novosti neft. i gaz. tekhn. Neftepererabotka i
neftekhimiya, no. 1, 1962, 9-13)

TEXT: Extracts obtained in the solvent refining of transformer distillates
yield a sulfurous fraction containing sulfides. It has been shown that
this fraction can be effective as an antioxidant for phenol-refined
transformer oils. Some individual sulfur compounds were tested: cyclo-
hexyl decyl sulfide and decyl thiophane were found to be efficient inhibitors.
Furfural-refined oils were found to contain the largest proportion of the sulfide sulfur (62-65% of the total sulfur content),
which appears to be one of the main factors of stability of these oils. When the sulfurous crude petroleum is used to prepare the

Card 1/2

Effects of sulfur compounds ...

S/081/62/000/021/047/069
B171/B101

transformer oil, the authors recommend to adjust the refining conditions
in order to leave ~ 0.5% of sulfide sulfur in the finished product.
[Abstracter's note: Complete translation.]

Card 2/2

L 9100-65

ENT(m)/EPF(c)/EWP(j)/T Pe-4/Pt-4 ASD(p)-3/RAEM(i) DJ/RM

ACCESSION NR: AT3001316

S/2933/63/005/000/0213/0218

AUTHOR: Krol', B. B.; L. G. Zherdeva; Z. I. Rozanova; A. A. Rozhdestvenskaya

TITLE: Effect of organic sulfur compounds on the stability of transformer oil from sulfur-containing crudes

SOURCE: AN SSSR. Bashkirskiy filial Khimiya seraorganicheskikh soyedineniy soderzhashchikhsya v neftyakh i nefteproduktakh, v. 5, 1963, 213-218

TOPIC TAGS: oil, transformer oil, organic sulfur, sulfoxide, oil stability, sulfurous crude oil, antioxidant, phenyldecyldisulfide, tetradecyldisulfide, fural refining, phenol refining

ABSTRACT: Narrow fractions of transformer oil containing organic sulfur compounds of different types, with a total sulfur concentration of 6-10% were investigated for anti-oxidant activity; these included fractions in which the organosulfur compounds consisted of 80-90% sulfide, fractions without sulfide sulfur, and fractions with sulfoxides. The effect of individual organosulfur additives with different structures on the susceptibility to oxidation was also investigated. The physico-chemical properties (density, birefringence, dispersion, molecular weight) of the additive fractions are tabulated. Stability data show that the natural sulfur compounds consisting mostly of sulfides are effective antioxidants.

1/3

Card

L 9100-65

7

ACCESSION NR: AT3001316

especially with respect to suppressing the formation of acid products in the initial stage of oxidation. Sulfoxides (thiophanes) are also effective antioxidants. The effect of synthetic organic sulfur compounds on the stability of the hydrocarbons was investigated on the same transformer oil samples. The addition of 0.5-1% decylthiophane, for example, increased the stability considerably, giving the same effect as the sulfoxide fraction. A decrease in the number of aromatic nuclei in the sulfide radical or the substitution of the aromatic nucleus by cyclohexyl increased the activity of the sulfide. Phenyldecyl sulfide and tetracyl decyl sulfide produced the same results. The most effective antioxidant was found to be cyclohexyldecyl sulfide, i.e. a sulfide with both aliphatic and naphthalene radicals. After adding 1% cyclohexyldecyl sulfide to the oil, no sediment was found in the oxidized oil, and the acid number was 0.05-0.09 mg KOH/g. The stability of transformer oil depending on the method of refining (furfural, phenol) and the content of natural and sulfide sulfur is shown by tabulated data. Oils with a high sulfide content are more stable than those containing small amounts of sulfide, such as oils obtained by phenol extraction. Furfural-refined oils containing a large amount (62-65%) of sulfide sulfur (calculated for total sulfur content) are very stable. "The organic sulfur compounds were synthesized at the Kafedra khimii nefti MGU im. M. V. Lomonosova (Department of Petroleum Chemistry, Moscow State University). Part of the experimental work was done by F. G. Sidlyaronok, Ye. V. Voznesenskaya, F. S. Yakobi and V. I. Kutukova." Orig. art. has 6 tables.

2/3

Caro

L 9100-65

ACCESSION NR: AT3001316

ASSOCIATION: Vsesoyuznyy nauchno-issledovatel'skiy institut po pererabotke nefti i gaza i poluchenii iskusstvennogo zhidkogo topliva (All-Union Scientific Research Institute for the Refining of Petroleum and Gas and the Preparation of Synthetic Liquid Fuel)

SUBMITTED: 00

ENCL: 00

SUB CODE: FP, OC

NO REF SOV: 006

OTHER: 003

3/3

Cord

L 10225-63

EPF(c)/EWT(m)/BDS--AFFTC/ASD/APGC--Pr-4--RM/EW/MAY/MN/WW

ACCESSION NR: AP3000502

S/0065/63/000/005/0026/0031

70

68

AUTHOR: Krol', B. B.; Rozanova, Z. I.; Rozhdestvenskaya, A. A.

TITLE: Study of fractions of sulfur compounds from the 300-400C distillate of Tuymazy petroleum //

SOURCE: Khimiya i tekhnologiya topliv i masel, no. 5, 1963, 26-31

TOPIC TAGS: sulfur compounds, 300-400C distillate, Tuymazy petroleum, sulfide, benzo-thiophenes, thiophenes, oxidation stability, transformer O.I., oxidation inhibitors, chromatography

ABSTRACT: Fractions of sulfur compounds were separated from extracts obtained in purifying distillate with phenol in the Novo-Ufimskiy NPZ. The extracts were dewaxed and resolved by chromatography on silica gel and aluminum oxide. Sulfides were separated from other sulfur compounds by oxidation with hydrogen peroxide. The narrow fractions obtained contained up to 10% S and had molecular weights in the range 230-260. Their physical and chemical properties indicate that the distillate studied contains benzo-thiophenes and thiophanes. The fractions isolated including sulfoxides were compared with synthetic sulfides and thiophenes in their effect on the oxidation stability of transformer O.I. Both natural and synthetic

Card 1/2

L 10225-63

ACCESSION NR: AP3000502

2

sulfides are oxidation inhibitors while benzethiophene derivatives are ineffective.
Orig. art. has: 1 flow diagram and 3 tables.

ASSOCIATION: None

SUBMITTED: 00

DATE ACQD: 12Jun63

ENCL: 00

SUB CODE: CH

NO REF Sov: 006

OTHER: 005

kes/mh
Card 2/2

KROL', B.B.; ROZANOVA, Z.I.; ROZHDESTVENSKAYA, A.A.

Study of the fractions of sulfur-containing compounds from
the Tuymazy petroleum distillate boiling at 300-400°C. Khim.
i tekhn. topl. i masel 8 no.5:26-31 My '63. (MIRA 16:8)

KROL', S.B.; ZHERDEVA, L.G.; IOGANSEN, V.; ROZANOVA, Z.I.

Composition and properties of aromatic hydrocarbons isolated
from the 300°-400° distillate of Tuy-mazy oil. Trudy VNII NP
no. 7:48-62. '58.
(Tuy-mazy--Petroleum) (Hydrocarbons)

S/081/62/000/004/071/087
B138/B110

AUTHORS: Krol', B. B., Zherdeva, L. G., Rozanova, Z. I.

TITLE: The influence of natural organosulfur compounds on the stability of low viscosity oil from Tuymazy crudes

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 4, 1962, 483, abstract 4M191 (Sb. "Khimiya seraorgan. soyedineniy, soderzhashchikhsya v neftyakh i nefteproduktakh. v. 4". M., Gostoptekhizdat, 1961, 189-193)

TEXT: In experiments carried out in connection with a study of the stability of low S deep refined transformer oil from Tuymazy crude, and also of the detarred part of this oil before and after removal of the S-compounds, the natural S-containing organic compounds were found to have a positive effect on the stability of the oil with regard to oxidation. The S-compounds appear to play a particularly positive role as oxidation inhibitors for oil consisting of naphthenic and monocyclic aromatic hydrocarbons. When the cyclicity of the aromatics contained in the oil is increased, the inhibiting effect of the S-compounds is maintained.
[Abstracter's note: Complete translation]

Card 1/1

10

15

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25

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"APPROVED FOR RELEASE: 07/19/2001

CIA-RDP86-00513R001445530008-1

PINCHUK, Mariya, zvenevaya; ROZANOVICH, Sasha [Razanovich, Sasna];
ZHOGAI', Matrena [Zhohai', Matruna]

A field crew of close friends. Rab. sial. 37 no.9:8-9 S '61.
(MIRA 14:10)

(Stolin District--Women as farmers)

APPROVED FOR RELEASE: 07/19/2001

CIA-RDP86-00513R001445530008-1"

PCZAV 1965 N. A.

USSR/Geology
Stratification

"Fundamental problems of Folding Mechanism," V. V. Belousov (with participation of I. V. Kirillovov, N. A. Rozanovov, A. V. Goryacheva), 26pp

"Byull Moskov Obsh Isp Pri, Nova Ser, Otdel Geol" Vol XXII, No 3

Kinematic mechanism of folding depends more on redistribution of plastic rocks than on harder interlayers. Because the flow of different rock layers varies in intensity, the material presses out to form the crest and troughs of a fold. The greater the plasticity, the deeper the fold. Such deformation produces cleavage of several types: main cleavage (parallel to axial surfaces of the fold); fanlike (converging along anticlines); the S-like (curving of main cleavage); dynamic (curving of separate layers); and transverse cleavage. Further decrease of the deformation plasticity and sliding concentration results in the formation of paraclastes and dioclases.

PA 49T28

"APPROVED FOR RELEASE: 07/19/2001

CIA-RDP86-00513R001445530008-1

ROZANOWICZ, Andrzej, dr

Crew fluctuation and the sickness and accident frequency in
coal mining. Przegl gorn 20 no. 5/230-232 Mj '64.

APPROVED FOR RELEASE: 07/19/2001

CIA-RDP86-00513R001445530008-1"

GLOWINSKI, Mieczyslaw; LAKOMY, Tadeusz; LIMANSKI, Marian; ROZANOWICZ, Andrzej

Blood serum electrophoregrams in normal women several weeks after the conception and their utilization in the study of pathological conditions in pregnancy. Gin. polska 32 no.4:423-441 '61.

1. Z Kliniki Poloznictwa i Chorob Kobiecych Slaskiej AM w Zabrzu. Kierownik: prof dr W. Starzewski. Z Zakladu Higieny Ogolnej Slaskiej AM w Zabrzu. Kierownik: prof. dr med. B. Nowakowski. Z Wojewodzkiej Przychodni Immunopatologii Ciazy i Noworodka w Katowicach. Kierownik: dr med. M. Skorczynski

(PREGNANCY blood)

(BLOOD PROTEINS in pregn)

ROZANOWSKI, Jerzy, mgr., inz.

The development os ship power stations. Tech gosp morska 11 no.7/8:
213-215 J1-Ag '61.

1. Centralne Biuro Konstrukcji Okretowych nr. 1, Gdansk.

ROZANOWSKI, Jerzy, mgr., inz.

The development of ship power stations. Tech gosp morska 11 no.7/8:
213-215 '61. (EEAI 10:9/10)

1. Centralne Biuro Konstrukcji Okretowych nr. 1, Gdansk.

(Electricity on ships)

ROZANOWSKI, Jerzy, mgr., inz.

Reduction of engine-room noise. Bud. okretowe Warszawa 6 no. 8:255-259
'61.

1. Centralne Biuro Konstrukcji Okretowych Nr 1, Gdansk.

(Ships) (Noise)

ROZANSKA, M.

GOSCZCZYNSKA, K., mgr.; ROZANSKA, M., mgr

Determination of pharmaceutic preparation with perchloric acid in
anhydrous medium. Farm. polska 10 no.6:153-156 June 54.

1. Z Zakladu Chemii Instytutu Lekow w Warszawie.

(DRUGS, determination

with perchloric acid, in anhydrous medium)

(ACIDS,

perchloric, determ. of drugs in anhydrous medium)

(CHLORINE,

perchloric acid, determ. of drugs in anhydrous medium)

JASTRZEBSKI, T.; ROZANSKA, M. (Lublin - Pulawy)

Attempts at titration of mallein by the intradermal method applied
on horses sensitized by killed glanders germs. Report I. Sensitiza-
tion of horses. Rocznauk roln wet 70 no.1/4:295-296 '60.
(EEAI 10:9)

(Glanders) (Mallein test)

POLAND

ROZANSKA, M., Department of Poultry Diseases, Veterinary Institute
(Zaklad Chorob Drobnych Instytutu Wet.), Pulawy. Prof. Dr. K. Marek, Head.

"Specificity of the Agglutination Reaction in the Detection of Tuberculosis
Bacilli Carriers in Poultry"

Lublin, Medycyna Weterynaryjna, Vol 22, No 4, 1966, pp 193-196.

Abstract: Serological tests were carried out on birds experimentally infected with tuberculosis, naturally infected with *S. pullorum* and immunized with killed *S. typhymurium* and *E. coli* to demonstrate the specificity of the agglutination reaction. No agglutination was observed with the antigen for detection of tuberculosis in the birds with antibodies against *S. pullorum*, *S. typhymurium* and *E. Coli*.

Contains a summary in English, 4 Tables and 4 references (1 Polish, 1 Russian, 1 Czechoslovak and 1 German-language).

ROZANSKA, Miroslawa (Pulawy)

Evaluation of the potency of PPD avian tuberculin. Rocznik nauk rolnictwa wet
70 no.1/4:162-163 '60. (EEAI 10:9)

(Tuberculin)

SZYPTER, Karol, mgr inż.; ROZANSKI, Edward

Recovery of insulators. Wiad elekrotechn 53 [i.e. 32] no.4:
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1. Electric Network Constructions, Katowice.

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NOBIS, Wladyslaw; ROZANSKI, Janusz

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against underwashing, based on model tests. Rozpr hydrotechn
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ROZANSKI, Lech

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1. Submitted December 1963.

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Vol. 10, No. 12, Dec. 1954. Stalinogrod, Poland)

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I. General Executive, Plenipotentiary Minister of Mining and Power Engineering for problems of water and sewage management, Warsaw.

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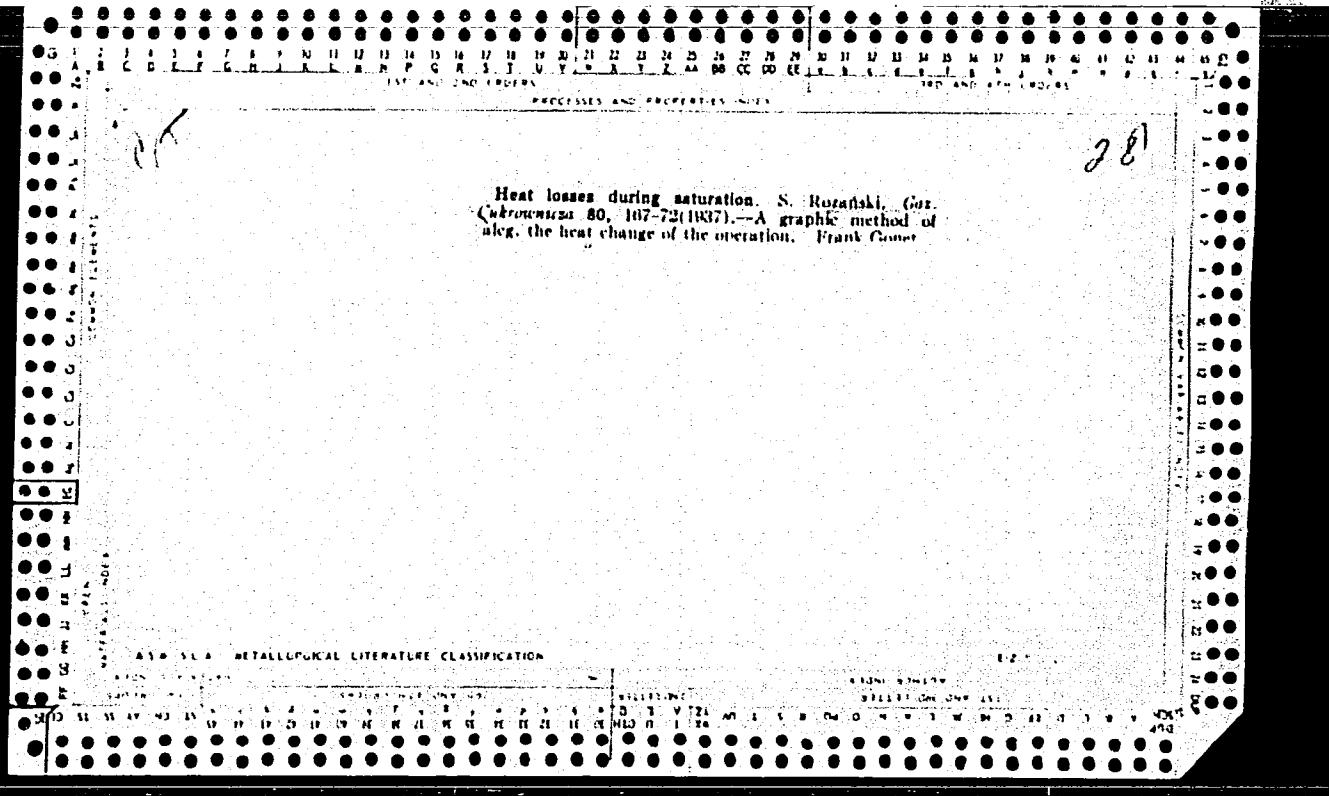
ROZANSKI, Stanislaw; TARAJKOWSKA, Mieczyslawa; ZYCH, Stanislaw

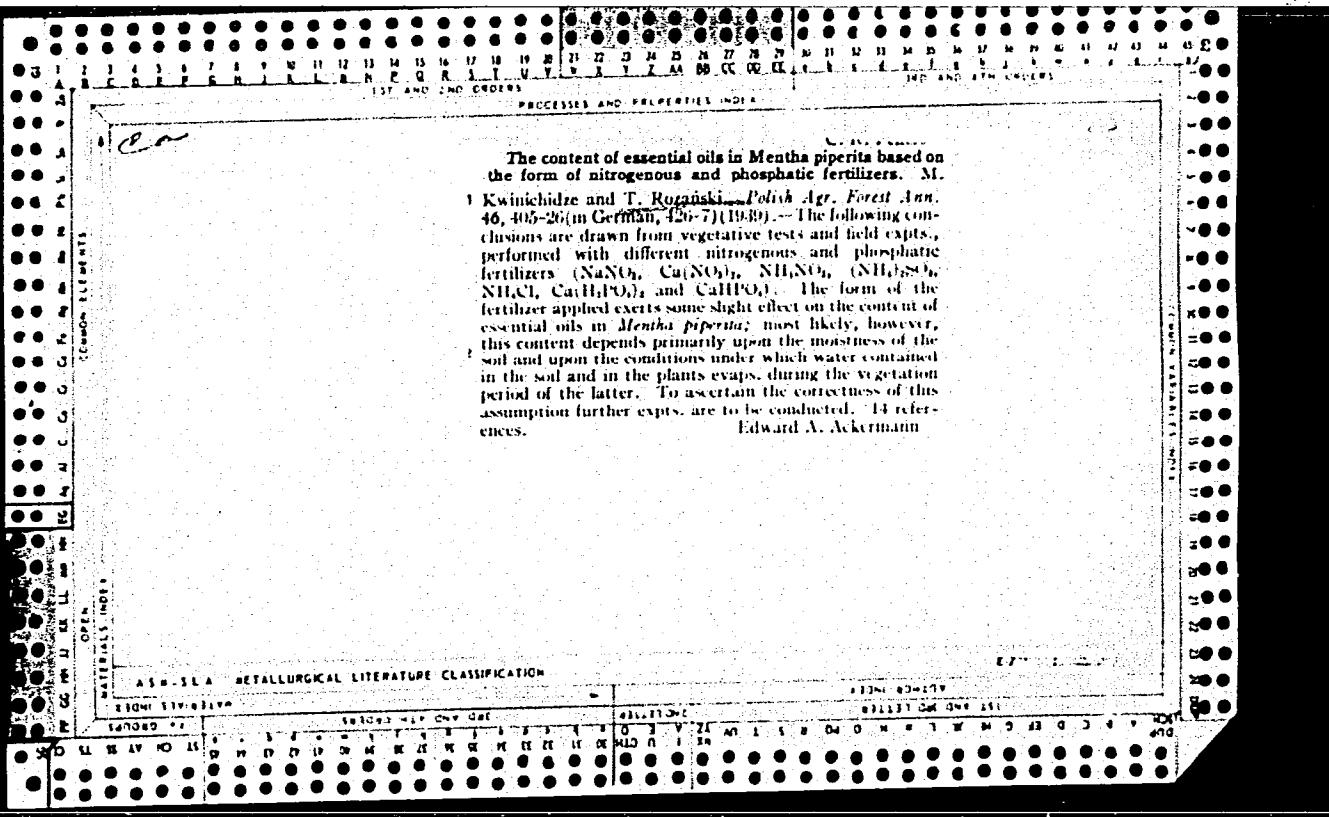
Some results of researches on the climate of Lodz.
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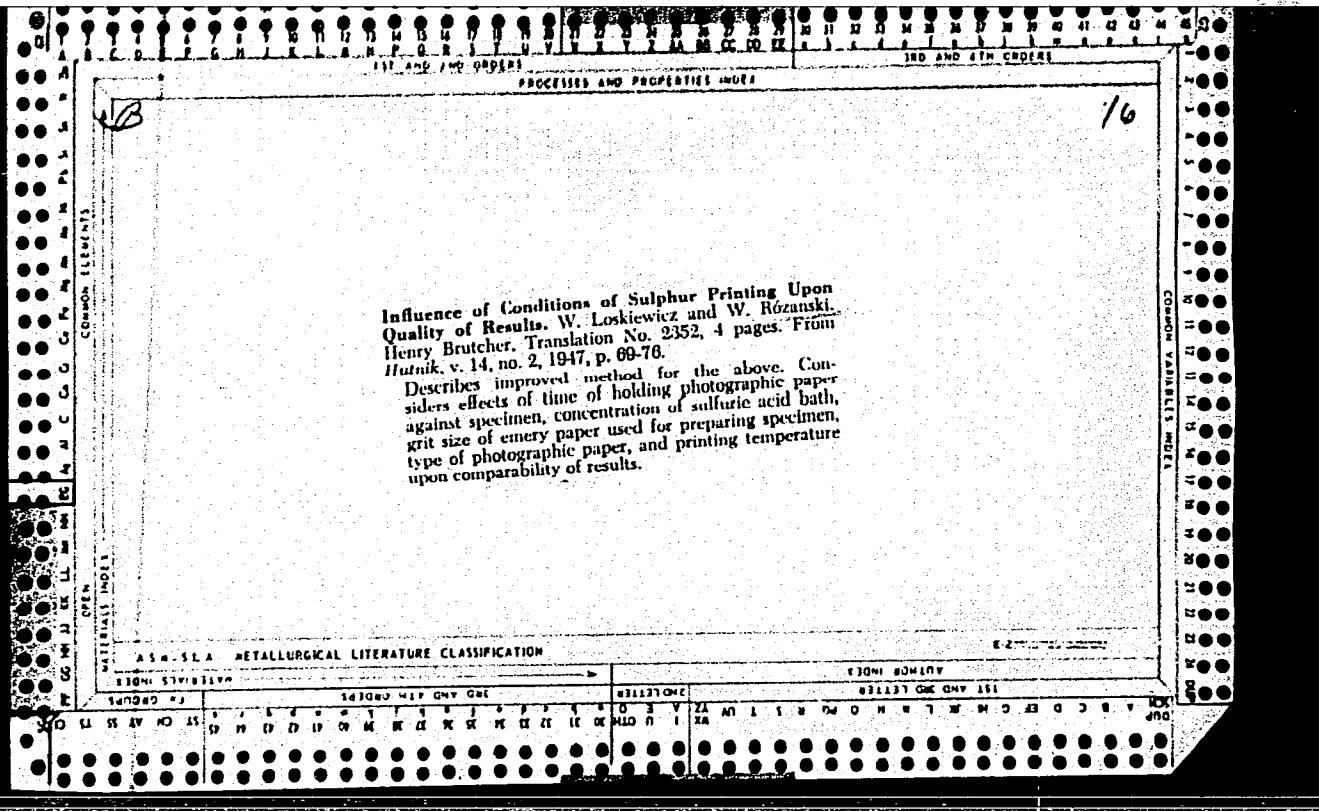
Przegl

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ROZANSKI, W., doc. dr inz.; RYS, J., dr inz.; KOSOWSKA, E., mgr inz.

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